

Stereoselective Crossed-Aldol Condensation of Hetaryl methyl Ketones with Aromatic Aldehydes in Water : Synthesis of (2E)-3-Aryl-1-hetarylprop-2-en-1-ones

SALEM A. BASAIF, TARIQ R. SOBAHI, ALI KH. KHALIL* and MOHAMED A. HASSAN*

*Chemistry Department, Faculty of Science, King Abdulaziz University,
Jeddah, Saudi Arabia.*

* *Chemistry Department, Faculty of Science, Ain Shams University, Abbassia, Cairo, Egypt*
E-mail : mahassan77@yahoo.com

ABSTRACT. Aldol condensation of 2-acetylthiophene, 2-acetylpyrrole and 2-acetylpyridine with different aromatic aldehydes were carried out in water in heterogeneous phases in the presence of cetyltrimethyl-ammonium bromide as cationic surfactant at room temperature. All the reactions occur in a short time with excellent yields of stereoselective hetarylpropanes in water as environmental friendly solvent.

Keywords : *Crossed-Aldol. Hetarylpropenones. Chalcones. Synthesis in water.*

Introduction

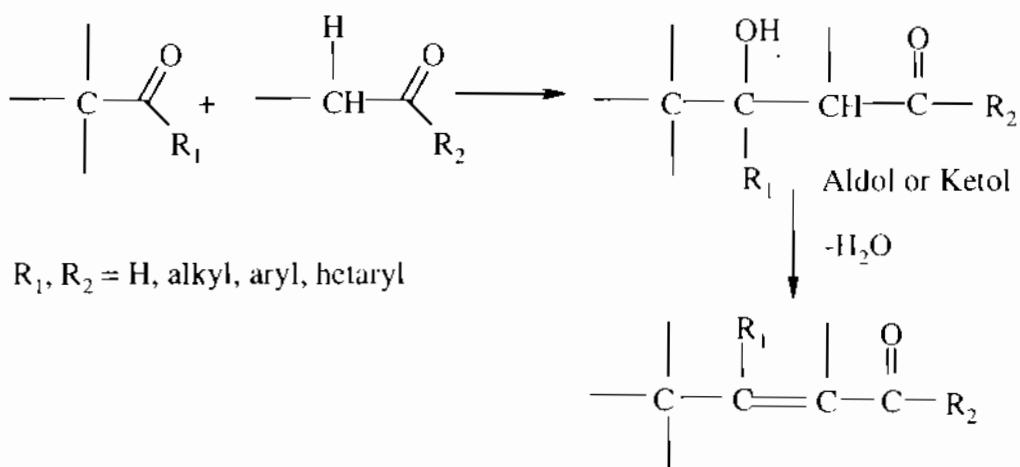
Chalcones are α,β -unsaturated ketones and they have great existence in the plant kingdom. It is well known that most of natural or synthetic chalcones are highly biologically active with a great pharmaceutical and medicinal applications^[1]. Recently they are used as anti-AIDS agents^[2], cytotoxic with antiangiogenic activity^[3,4], antimalarial^[5,6], anti-inflammatory^[7,8] and antitumor^[9,10].

The U.S. Environmental Protection Agency (EPA) has suggested a drastic reduction of using of more than ten of hazard common organic solvents in the industrial production of chemicals. We are dealing in this paper a cleaner and safe production of high yield of stereoselective hetarylchalcones as an important biologically active compounds in water as a cheap solvent and environmental friendly reaction medium.

Recently, water is an attractive medium for many organic reactions^[11]. The important advantageous of aqueous medium with respect to organic solvents are less expensive, healthy, safe and environmentally friendly. Also, it allows the pH control and the use of surfactants as micro aggregates.

The hydrophobic effect and the large cohesive energy of water are considered to be the main factors responsible for increasing reactivity and selectivity of the reactions^[12,13].

Mixed or crossed aldol condensation is a base-catalyzed addition of different aldehydes and ketones, one of them must contain at least one α -hydrogen to give an aldol or ketol which are dehydrated to α,β -unsaturated aldehydes or ketones.



The classical reaction conditions of aldol condensation are NaOH solution in hydroalcoholic medium which are, often, yielded a mixture of (*E*) and (*Z*) chalcones^[14,15].

Recently, aldol reaction can, also, be carried out in an aqueous medium in the presence of catalysts to increase molecular aggregations and stereoselectivity^[16-19]. It is considered cleaner conditions of the production of some known and unknown chalcones.

Experimental Section

All melting points reported are uncorrected. IR spectra were recorded using Perkin Elmer's Spectrum RXIFT-IR spectrophotometer (ν in cm^{-1}) using KBr. The NMR spectra were recorded on Bruker Avance DPX400 spectrometer, using CDCl_3 as solvent and TMS as internal standard (chemical shifts in δ values in ppm, J in Hz). Elemental analyses were performed on Perkin Elmer 2400, series II microanalyzer. Fine chemicals are Aldrich products and they are used without further purification.

General procedure

Hetarylmethyl ketones (1, 3, 5, 100 mmol), aromatic aldehydes (100 mmol) and cetyl trimethylammonium bromide (CTABr) (5.46g, 15 mmol) were added to an aqueous solution of NaOH (200 ml, 0.5 M). The mixture was vigorously stirred at 20 °C for the time reported in Tables 1, 2 and 3. The reaction was monitored by TLC of dissolving sample of reaction mixture in CH_2Cl_2 during the reaction period using ether : pet.ether (1:3) as eluent. The solid products were filtered off, washed with water (3x25 ml), dried and crystallized from the proper solvent. The yields of the purified specimens are listed in Tables 1, 2 and 3.

The general procedure offers the following significant advantages over conventional procedures:

- 1- Improved reaction rates and increased yields through suppression of side reactions.
- 2- Clean, safe, and simple methodology.
- 3- Modifications of stereo-selectivity.
- 4- No need for expensive and hazard organic solvents.

- 5- Aqueous alkali metal hydroxides replace alkoxides.
- 6- Lower reaction temperatures and easier work-up.

(2E)-3-Phenyl-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2a): Pale yellow crystals from methanol; m.p. 136-138 °C; C₁₃H₁₁NO (197.24); calcd.: C, 79.16; H, 5.62; N, 7.10; found: C, 79.07; H, 5.54; N, 7.13. IR: 1642 (C=O), 2851, 2918, 3026 (CH), 3268 (NH). ¹H-NMR: δ; 6.38 (m, 1H), 7.07 (d, 1H), 7.18 (d, 1H), 7.32 (d, 1H, C₂-H; J=15.60), 7.51-7.65 (m, 5H), 7.85 (d, 1H, C₃-H; J=15.58), 10.32 (br, 1H, NH, D₂O exchangeable).

(2E)-1-(1H-pyrrol-2'-yl)-3-(4'-tolyl)prop-2-en-1-one (2b): Pale yellow crystals from ethanol; m.p. 148-150 °C; C₁₄H₁₃NO (211.26); calcd.: C, 79.60; H, 6.20; N, 6.63; found: C, 79.51; H, 6.13; N, 6.57. IR: 1642 (C=O), 2995 (CH), 3257 (NH). ¹H-NMR: δ; 2.41(s, 3H), 6.37 (m, 1H), 7.12 (d, 1H), 7.23 (s, 1H), 7.27 (d, 2H), 7.36 (d, 1H, C₂-H, J=15.62), 7.56 (d, 2H), 7.85 (d, 1H, C₃-H; J=15.68), 10.34 (br, 1H, NH, D₂O exchangeable).

(2E)-3-(2'-Chlorophenyl)-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2c): Pale green crystals from ethanol; m.p. 121-123 °C; C₁₃H₁₀ClNO (231.68); calcd.: C, 67.40; H, 4.35; N, 6.05; found: C, 67.32; H, 4.30; N, 5.97. IR: 1645 (C=O), 2874, 2986 (CH), 3274 (NH). ¹H-NMR: δ; 6.37 (m, 1H), 6.93 (d, 1H), 7.10 (s, 1H), 7.25 (s, 1H), 7.27 (d, 1H), 7.32 (d, 1H, C₂-H; J=15.69), 7.44 (d, 1H), 7.76 (t, 1H), 8.23 (d, 1H, C₃-H; J=15.69), 10.12 (br, 1H, NH, D₂O exchangeable).

(2E)-3-(4'-Chlorophenyl)-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2d): Pale green crystals from ethanol; m.p. 154-156 °C; C₁₃H₁₀ClNO (231.68); calcd.: C, 67.40; H, 4.35; N, 6.05; found: C, 67.32; H, 4.30; N, 5.97. IR: 1645 (C=O), 2874, 2980 (CH), 3274 (NH). ¹H-NMR: δ; 6.36 (m, 1H), 7.10 (s, 1H), 7.16 (s, 1H), 7.36 (d, 1H, C₂-H; J=15.78), 7.38 (d, 2H), 7.56 (d, 2H), 7.78 (d, 1H, C₃-H; J=15.72), 10.34 (br, 1H, NH, D₂O exchangeable). ¹³C-NMR: δ; 111.32 (CH), 117.11 (CH), 122.80 (CH), 124.40 (CH), 128.11 (CH), 129.63 (C), 131.38 (2xCH), 134.19 (C₂-H), 134.65 (C), 137.22 (C), 142.31 (C₃-H), 179.77 (C=O).

(2E)-3-(2'-Hydroxyphenyl)-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2e): Yellow crystals from methanol; m.p. 167-168 °C; C₁₃H₁₁NO₂ (213.23); calcd.: C, 73.23; H, 5.20; N, 6.57; found: C, 73.11; H, 5.14; N, 6.50. IR: 1632 (C=O), 2853, 2919 (CH), 3258 (NH), 3453 (OH). ¹H-NMR: δ; 6.25 (m, 1H), 6.59 (m, 1H), 6.86 (d, 1H), 7.08 (m, 1H), 7.11 (d, 1H), 7.27 (d, 1H, C₂-H; J=15.58), 7.38 (d, 1H), 7.63 (m, 1H), 8.14 (d, 1H, C₃-H; J=15.66), 10.04 (br, 1H, NH, D₂O exchangeable), 10.33 (br, 1H, NH, D₂O exchangeable).

(2E)-3-(4'-Methoxyphenyl)-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2f): Pale yellow crystals from pet. ether 60-80 °C; m.p. 135-137 °C; C₁₄H₁₃NO₂ (227.26); calcd.: C, 73.99; H, 5.77; N, 6.16; found: C, 73.86; H, 5.71; N, 6.05. IR: 1641 (C=O), 2842, 2970 (CH), 3259 (NH). ¹H-NMR: δ; 3.85 (s, 3H), 6.36 (m, 1H), 6.94 (d, 2H), 7.10 (d, 1H), 7.15 (d, 1H), 7.28 (d, 1H, C₂-H; J=15.62), 7.61 (d, 2H), 7.83 (d, 1H, C₃-H; J=15.59), 10.46 (br, 1H, NH, D₂O exchangeable). ¹³C-NMR: δ; 57.86 (CH₃), 111.72 (CH), 115.31 (2xCH), 117.10 (CH), 118.94 (C), 121.63 (CH), 126.42 (C), 130.20 (2xCH), 131.73 (C₂-H), 143.84 (C₃-H), 163.21 (C), 180.90 (C=O).

(2E)-3-(3',4'-Methylenedioxyphenyl)-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2g):

Pale yellow crystals from ethanol; m.p. 140-142 °C; $C_{14}H_{11}NO_3$ (241.42); calcd.: C, 69.65; H, 4.59; N, 5.80; found: C, 69.57; H, 4.53; N, 5.71. IR: 1642 (C=O), 2830, 2962 (CH), 3223 (NH). 1H -NMR: δ ; 6.06 (s, 2H), 6.37 (m, 1H), 6.85 (d, 1H), 7.09-7.15 (m, 3H), 7.17 (d, 1H), 7.23 (d, 1H, C_2 -H; J =15.63), 7.97 (d, 1H, C_3 -H; J =15.62), 10.49 (br, 1H, NH, D₂O exchangeable). ^{13}C -NMR: δ ; 103.27 (CH₂), 105.65 (CH), 107.63 (CH), 109.85 (CH), 116.22 (CH), 119.48 (C₂-H), 124.23 (CH), 129.45 (C), 133.11 (C), 142.53 (C₃-H), 148.27 (C), 148.33 (C), 178.84 (C=O).

(2E)-3-(4'-N,N-dimethylaminophenyl)-1-(1H-pyrrol-2'-yl)prop-2-en-1-one (2h):

Deep yellow crystals from methanol; m.p. 192-194 °C; $C_{15}H_{16}N_2O$ (240.30); calcd.: C, 74.98; H, 6.71; N, 11.58; found: C, 74.92; H, 6.65; N, 11.53. IR: 1638 (C=O), 2907 (CH), 3241 (NH). 1H -NMR: δ ; 3.04 (s, 6H), 6.34 (m, 1H), 6.70 (d, 2H), 7.05 (m, 2H), 7.18 (d, 1H, C_2 -H; J =15.58), 7.54 (d, 2H), 7.81 (d, 1H, C_3 -H; J =15.53), 10.38 (br, 1H, NH, D₂O exchangeable).

(2E)-1-(Thien-2'-yl)-3-(4'-tolyl)prop-2-en-1-one (4a): Pale yellow crystals from ethanol; m.p. 112-114 °C; $C_{14}H_{12}OS$ (228.31); calcd.: C, 73.65; H, 5.30; found: C, 73.58; H, 5.27. IR: 1589 (C=C), 1647 (C=O), 2918, 3083 (CH). 1H -NMR: δ ; 2.40 (s, 3H), 7.18 (m, 1H), 7.21 (d, 2H), 7.39 (d, 1H, C_2 -H; J =15.65), 7.55 (d, 2H), 7.67 (d, 1H), 7.83 (d, 1H, C_3 -H; J =16.13), 7.87 (d, 1H). ^{13}C -NMR: δ ; 21.59 (CH₃), 120.59 (CII), 128.25 (CH), 128.56 (2xCH), 129.74 (2xCH), 131.71 (C₂-H), 131.97 (C), 133.78 (CH), 141.21 (C), 144.21 (C₃-H), 145.68 (C), 182.18 (C=O).

(2E)-3-(4'-Chlorophenyl)-1-(thien-2'-yl)prop-2-en-1-one (4b): Yellow crystals from ethanol; m.p. 118-120 °C; $C_{13}H_9ClOS$ (248.73); calcd.: C, 62.78; H, 3.65; found: C, 62.66; H, 3.62. IR: 1591 (C=C), 1645 (C=O), 3089 (CH). 1H -NMR: δ ; 7.20 (m, 1H), 7.23 (d, 2H), 7.41 (d, 1H, C_2 -H; J =15.65), 7.64 (d, 2H), 7.72 (d, 1H), 7.85 (d, 1H, C_3 -H; J =15.79), 7.93 (d, 1H). ^{13}C -NMR: δ ; 121.96 (CH), 128.28 (CH), 129.22 (2xCII), 129.60 (2xCH), 131.90 (C₂-H), 133.13 (C), 134.11 (CH), 136.45 (C), 142.55 (C₃-H), 145.31 (C), 181.74 (C=O).

(2E)-3-(4'-BromoPhenyl)-1-(thien-2'-yl)prop-2-en-1-one (4c): Pale yellow crystals from ethanol; m.p. 131-133 °C; $C_{13}H_9BrOS$ (293.18); calcd.: C, 53.26; H, 3.09; found: C, 53.19; H, 3.05. IR: 1581 (C=C), 1649 (C=O), 2903-3085 (CH). 1H -NMR: δ ; 7.21 (m, 1H), 7.48 (d, 1H, C_2 -H; J =15.58), 7.52 (d, 2H), 7.57 (d, 2H), 7.72 (d, 1H), 7.80 (d, 1H, C_3 -H; J =15.58), 7.89 (d, 1H). ^{13}C -NMR: δ ; 122.07 (CH), 124.84 (C), 128.29 (CH), 128.77 (2xCH), 131.93 (C₂-H), 132.18 (2xCH), 133.56 (C), 134.13 (CH), 142.62 (C₃-H), 145.30 (C), 181.72 (C=O).

(2E)-3-(4'-Methoxyphenyl)-1-(thien-2'-yl)prop-2-en-1-one (4d): Yellow crystals from ethanol; m.p. 144 -146 °C; $C_{14}H_{12}O_2S$ (244.31); calcd.: C, 68.83; H, 4.95; found: C, 68.77; H, 4.90. IR: 1590 (C=C), 1647 (C=O), 2838-3082 (CH). 1H -NMR: δ ; 3.83 (s, 3H), 6.96 (d, 2H), 7.24 (m, 1H), 7.32 (d, 1H, C_2 -H; J =15.62), 7.38 (d, 1H), 7.65 (d, 2H), 7.72 (d, 1H), 7.84 (d, 1H, C_3 -H; J =15.62).

(2E)-3-(2',4'-Dimethoxyphenyl)-1-(thien-2'-yl)prop-2-en-1-one (4e): Yellow crystals from ethanol; m.p. 113-115 °C; C₁₅H₁₄O₃S (274.34); calcd.: C, 65.67; H, 5.14; found: C, 65.58; H, 5.07. IR: 1564 (C=C), 1635 (C=O), 2839-3090 (CH). ¹H-NMR: δ; 3.85 (s, 3H), 3.90 (s, 3H), 6.52 (d, 2H), 7.15 (m, 1H), 7.45 (d, 1H, C₂-H; J=15.62), 7.55 (d, 1H), 7.63 (m, 1H), 7.83 (s, 1H), 8.07 (d, 1H, C₃-H; J=15.62).

(2E)-3-(3',4'-Dimethoxyphenyl)-1-(thien-2'-yl)prop-2-en-1-one (4f): Yellow crystals from ethanol; m.p. 119-121 °C; C₁₅H₁₄O₃S (274.34); calcd.: C, 65.67; H, 5.14; found: C, 65.61; H, 5.10. IR: 1578 (C=C), 1647 (C=O), 2847-3105 (CH). ¹H-NMR: δ; 3.88 (s, 3H), 3.92 (s, 3H), 6.83 (d, 1H), 6.99 (s, 1H), 7.13 (d, 1H), 7.18 (m, 1H), 7.34 (d, 1H, C₂-H; J=15.64), 7.58 (d, 1H), 7.65 (d, 1H, C₃-H; J=15.64), 7.68 (d, 1H).

(2E)-3-(3',4'-Methylenedioxyphenyl)-1-(thien-2'-yl)prop-2-en-1-one (4g): Pale yellow crystals from ethanol; m.p. 117-119 °C; C₁₄H₁₀O₃S (258.30); calcd.: C, 65.10; H, 3.90; S, 12.41; found: C, 64.96; H, 3.85. IR: 1587 (C=C), 1645 (C=O), 2906-3108 (CH). ¹H-NMR: δ; 6.07 (s, 2H), 6.76 (d, 1H), 7.15 (d, 1H), 7.22 (m, 2H), 7.32 (d, 1H, C₂-H; J=15.62), 7.63 (d, 1H), 7.78 (d, 1H, C₃-H; J=15.69), 7.86 (d, 1H). ¹³C-NMR: δ; 101.63 (CH₂), 106.62 (CH), 108.67 (CH), 119.84 (CH), 125.34 (C₂-H), 128.19 (CH), 129.19 (C), 131.54 (CH), 133.66 (CH), 143.90 (C₃-H), 145.65 (C), 148.36 (C), 149.94 (C), 181.93 (C=O).

(2E)-1,3-Di-(thien-2'-yl)prop-2-en-1-one (4h): Orange crystals from ethanol; m.p. 136-138 °C; C₁₁H₈OS₂ (220.32); calcd.: C, 59.97; H, 6.10; found: C, 59.91; H, 6.07. IR: 1572 (C=C), 1639 (C=O), 3092 (CH). ¹H-NMR: δ; 7.09 (d, 1H), 7.17 (m, 1H), 7.23 (d, 1H, C₂-H; J=15.72), 7.37 (s, 1H), 7.42 (m, 1H), 7.72 (m, 1H), 7.88 (s, 1H), 8.03 (d, 1H, C₃-H; J=15.63). ¹³C-NMR: δ; 120.34 (CH), 128.21 (CH), 128.35 (CH), 128.85 (C₂-H), 131.65(C₃-H), 132.20 (CH), 133.84 (CH), 136.46 (CH), 141.48 (C), 149.97 (C), 181.57 (C=O).

(2E)-3-(4'-Chlorophenyl)-1-(pyrid-2'-yl)prop-2-en-1-one (6a): Pale yellow crystals from ethanol; m.p. 103-105 °C; C₁₄H₁₀ClNO (243.69); calcd.: C, 69.00; H, 4.14; N, 5.75; found: C, 68.91; H, 4.09; N, 5.70. IR: 1568, 1606 (C=C, C=N), 1673 (C=O), 3020-3081 (CH). ¹H-NMR: δ; 7.38 (d, 2H), 7.50 (m, 1H), 7.66 (d, 2H), 7.87 (d, 1H, C₂-H; J=16.15), 7.88 (d, 1H), 8.19 (d, 1H), 8.28 (d, 1H, C₃-H; J=16.15), 8.74 (d, 1H). ¹³C-NMR: δ; 121.29 (CH), 122.96 (CH), 127.10 (C₂-H), 129.14 (CH), 129.96 (CH), 133.63 (C), 136.42 (C), 137.08 (C₃-H), 143.19 (CH), 148.87 (CH), 154.00 (C), 189.27 (C=O).

(2E)-3-(4'-Methoxyphenyl)-1-(pyrid-2'-yl)prop-2-en-1-one (6b): Pale yellow crystals from ethanol; m.p. 120-122 °C; C₁₅H₁₃NO₂ (239.27); calcd.: C, 75.30; H, 5.48; N, 5.85; found: C, 75.22; H, 5.43; N, 5.76. IR: 1570, 1596 (C=C, C=N), 1666 (C=O), 2845-3052 (CH). ¹H-NMR: δ; 3.86 (s, 3H), 6.94 (d, 2H), 7.48 (m, 1H), 7.69 (d, 2H), 7.86 (m, 1H), 7.92 (d, 1H, C₂-H; J=15.95), 8.17 (d, 1H, C₃-H; J=15.89), 8.19 (d, 1H), 8.74 (d, 1H). ¹³C-NMR: δ; 55.40 (CH₃), 114.32 (CH), 118.48 (CH), 122.87 (CH), 126.74 (CH), 126.94 (C₂-H), 127.94 (C), 130.67 (CH), 136.99 (C₃-H), 144.71 (CH), 148.79 (CH), 154.45 (C), 161.73 (C), 189.40 (C=O).

(2E)-3-(3',4'-Methylenedioxyphenyl)-1-(pyrid-2'-yl)prop-2-en-1-one(6c): Yellowish green crystals from ethanol; m.p. 148-150 °C; $C_{15}H_{11}NO_3$ (253.26); calcd.: C, 71.14; H, 4.38; N, 5.53; found: C, 71.08; H, 4.34; N, 5.47. IR: 1583 (C=C, C=N), 1656 (C=O), 2905-3054 (CH). 1H -NMR: δ ; 6.03 (s, 2H), 6.84 (d, 1H), 7.20 (d, 1H), 7.29 (s, 1H), 7.49 (m, 1H), 7.85 (d, 1H), 7.87 (d, 1H, C_2 -H; $J=15.88$), 8.06 (d, 1H, C_3 -H; $J=15.93$), 8.17 (m, 1H), 8.74 (d, 1H). ^{13}C -NMR: δ ; 101.62 (CH₂), 107.08 (CH), 108.59 (CH), 111.85 (CH), 118.89 (CH), 122.88 (CH), 125.68 (CH), 126.80 (C₂-H), 129.72 (C), 137.00 (C₃-H), 144.68 (CH), 148.37 (C), 148.82 (CH), 149.98 (C), 154.35 (C), 189.33 (C=O).

(2E)-1-(Pyrid-2'-yl)-3-(1H-pyrrol-2'-yl)prop-2-en-1-one (6d): Orange crystals from ethanol; m.p. 124-126 °C; $C_{12}H_{10}N_2O$ (198.22); calcd.: C, 72.71; H, 5.08; N, 14.13; found: C, 72.64; H, 5.06 N, 14.08. IR: 1573 (C=C, C=N), 1654 (C=O), 2998-3112 (CH), 3304 (NH). 1H -NMR: δ ; 6.34 (m, 1H), 6.73 (d, 1H), 7.00 (d, 1H), 7.26 (s, 1H), 7.48 (m, 1H), 7.85 (d, 1H, C_2 -H; $J=16.26$), 7.88 (m, 1H), 8.19 (d, 1H, C_3 -H; $J=16.28$), 8.71 (d, 1H), 8.88 (s, 1H). ^{13}C -NMR: δ ; 111.46 (CH), 113.94 (CH), 116.82 (CH), 122.79 (CH), 123.29 (CH), 126.60 (C₂-H), 134.14(CH), 136.28 (C), 137.05 (C₃-H), 148.57 (CH), 153.93 (C), 193.22 (C=O).

(2E)-1-(Pyrid-2'-yl)-3-(2'-thienyl)prop-2-en-1-one (6e): Pale green crystals from ethanol; m.p. 76-78 °C; $C_{12}H_9NOS$ (215.28); calcd.: C, 69.51; H, 4.21; N, 6.51; found: C, 69.47; H, 4.19; N, 6.43. IR: 1582 (C=C, C=N), 1665 (C=O), 3010, 3061 (CH), 3316 (NH). 1H -NMR: δ ; 7.08 (m, 1H), 7.45 (m, 2H), 7.49 (m, 1H), 7.86 (d, 1H, C_2 -H; $J=15.54$), 8.07 (s, 2H), 8.17 (d, 1H, C_3 -H; $J=15.52$), 8.74 (d, 1H). ^{13}C -NMR: δ ; 119.76 (CH), 122.81 (CH), 126.84 (C₂-H), 128.24 (CH), 129.18 (CH), 132.17 (CH), 136.97 (C₃-H), 137.22 (CH), 140.99 (C), 148.82 (CH), 155.32 (C), 193.21 (C=O).

Results and Discussion

We extended the previous investigations^[16-19] to carbon-carbon bond formation and we focus in this paper on crossed-alcohol condensation of some hetaryl methyl ketones with a variety of different aromatic aldehydes in water at room temperature and in the presence of cetyltrimethylammonium bromide (CTABr) as the proper cationic surfactant for the synthesis of (2E)-3-aryl-1-hetarylprop-2-en-1-ones in an excellent yields with high stereoselectivity. Analytical gas chromatography proved that, only, *E*-isomers of isolated propenones were detected. The 1H -NMR coupling constants (J) of C₂-H and C₃-H of the isolated hetarylpropenones are in the range of 15.5-16.0 Hz which are characteristic to *E*-propenones.

We are expected that the synthesized hetarylpropenones might have a biological and medicinal activities in analogy to the biologically active amino chalcones^[19], quinolinyl chalcones and some ferroenyl chalcone^[5].

Efficient stirring of an equimolar amount of 2-hetaryl methyl ketones (1, 3, 5) and aromatic aldehydes in aqueous NaOH solution and in the presence of cetyltrimethylammonium bromide (CTABr) as surfactant at room temperature, underwent stereoselective crossed-alcohol condensation with precipitation of the 3-aryl-1-hetarylprop-2-en-1-ones in high yields within a short reaction time (t) as shown in Tables 1, 2 and 3.

Table 1. Crossed-Aldol condensation of 2-acetylpyrrol (**1**) with aromatic aldehydes. Synthesis of (2E)-3-aryl-1-(pyrrol-2'-yl)prop-2-en-1-ones (**2a-h**).

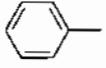
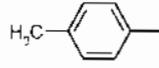
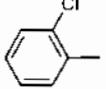
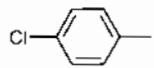
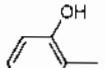
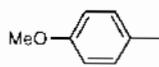
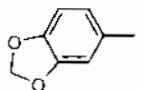
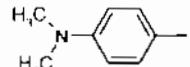
| Product No. | Ar | t (min) | Yield (%) |
|-------------|---|---------|-----------|
| 2a |  | 120 | 65 |
| 2b |  | 140 | 73 |
| 2c |  | 120 | 80 |
| 2d |  | 90 | 82 |
| 2e |  | 160 | 62 |
| 2f |  | 30 | 88 |
| 2g |  | 20 | 96 |
| 2h |  | 20 | 92 |

Table 2. Crossed-Aldol condensation of 2-acetylthiophene (**3**) with aromatic aldehydes. Synthesis of (*2E*)-3-aryl-1-(thien-2'-yl)prop-2-en-1-ones (**4a-h**).

| Product No. | Ar | t (min) | Yield (%) |
|-------------|----|---------|-----------|
| 4a | | 130 | 72 |
| 4b | | 100 | 76 |
| 4c | | 100 | 75 |
| 4d | | 45 | 83 |
| 4e | | 20 | 92 |
| 4f | | 30 | 84 |
| 4g | | 30 | 86 |
| 4h | | 100 | 74 |

Table 3. Crossed-Aldol condensation of 2-acetylpyridine (**5**) with aromatic aldehydes. Synthesis of (*2E*)-3-aryl-1-(pyrid-2'-yl)prop-2-en-1-ones (**6a-e**).

| Product No. | Ar | t (min) | Yield (%) |
|-------------|----|---------|-----------|
| 6a | | 40 | 88 |
| 6b | | 45 | 92 |
| 6c | | 20 | 95 |
| 6d | | 80 | 77 |
| 6e | | 60 | 82 |

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تكاثف الدول المختلط الانتقائي للأدヒدات الأروماتية مع الكيتونات المحتوية على حلقة غير متجانسة في الوسط المائي

سالم أحمد باسيف، و طارق رشاد سبجي، و محمد علي حسن*، و علي خليل علي *

قسم الكيمياء ، كلية العلوم ، جامعة الملك عبد العزيز

جدة - المملكة العربية السعودية

* قسم الكيمياء ، كلية العلوم ، جامعة عين شمس ، العباسية

القاهرة - مصر

المستخلص. تكاثفت الأدھيدات الأروماتية مع كل من ٢-أستيل ثيوفين ، ٢-أستيل بيرول وكذلك ٢-أستيل بيريدين في وجود بروميد سينيل رباعي مثل أمونيوم كمخفض للتوتر السطحي في الماء لتعطي نواتج تكاثف الدول وبابنتقائية مرتفعة ومردود عالي في زمن قصير.